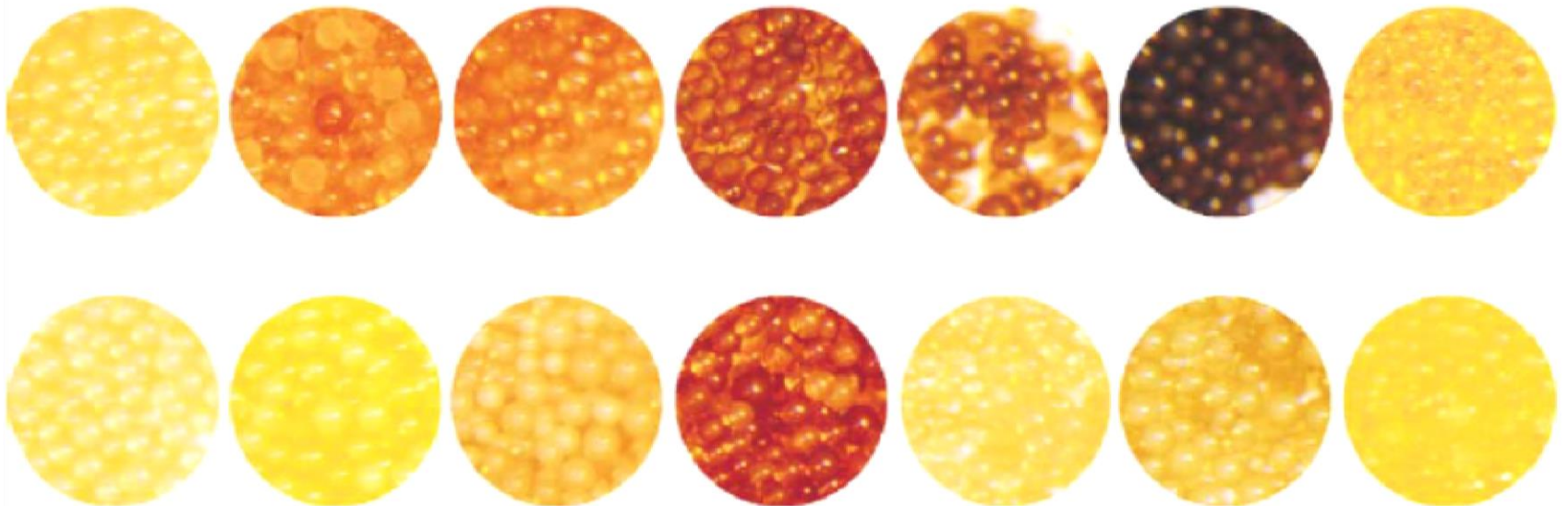


# Heterogenisierung von Homogenkatalysatoren durch organische Trägermaterialien – Design, (Kombinatorische) Synthesen und Anwendungen

Stefan Bräse

Institute of Organic Chemistry & Institute of Toxicology and Genetics



# Chemistry of the Bräse group: Functionalities and Function

## Drug delivery

*Bioconjugation*  
*Peptoids*



Organometallic  
Chemistry and  
Catalysis

## Bioactive entities

*Natural products*  
*Medicinal chemistry*  
*Combinatorial chemistry*  
*Pesticides*

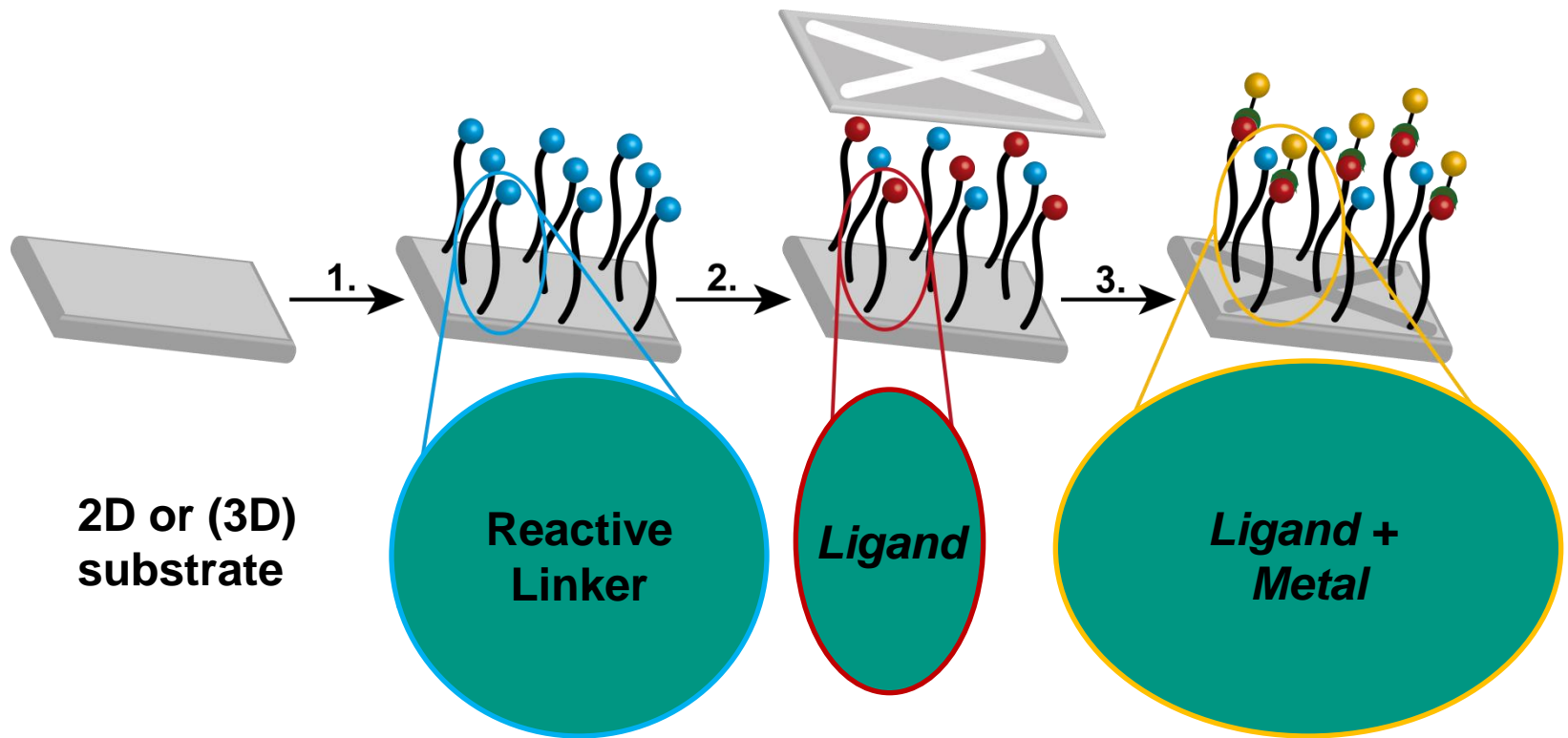
**Organic Synthesis  
of Functionalized  
Organic Structures**

Material Chemistry  
Nanostructures

# Today's menu

- **Aperitif: Immobilization of Homogeneous Catalysts**
- **Amuse-bouche: Asymmetric Catalysis**
- **Entrée: Immobilization of Catalysts**
- **Main course: Combinatorial Chemistry**
- **Dessert: Diversity through Multifunctional Linker**
- **Cheese: Auto-Click Reactions**

# Immobilization of Homogenous Catalysts Prinziples and Opportunities



# Immobilization of Homogenous Catalysts

## Pro's and Con's

### Pro's:

- Many designed ligands available
- Optimization of catalysts possible
- Asymmetric catalysis
- Reversible binding



### Con's:

- Metal leaching
- Time/space
- Lower catalyst loading
- Bimetallic catalysis/alloys difficult

# Immobilization of Homogenous Catalysts Suitable Substrates

- Polymers
  - Cross-linked polymers (e.g. Merrifield-type)
  - Hyper-cross-linked polymers and related systems
  - New: SURMOF and SURGELS
  - Nanoparticles for nanofiltration
  
- Hybrid system
- Inorganic substrates

# Three-dimensional polymers („Organic Frameworks“)



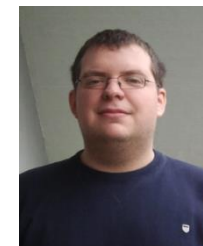
- The rigid structures are able to form
  - Hydrogen-bonded molecular networks
  - **Covalent-Organic Frameworks (COFs)**
  - **Hyper-Crosslinked Polymers (HCPs)**
  - **Surface-mounted Metal Organic Frameworks (SURMOFs, with C. Wöll)**
- Organic frameworks have high porosity and large surfaces
  - Specific Surface Area > 500 m<sup>2</sup>/g
- Applications
  - Gas storage and gas separation like zeolites
  - Bio-applications
  - **Catalysis**



A. Schade



I. Wessely

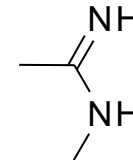
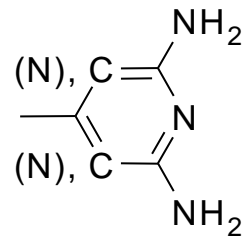
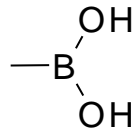
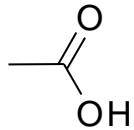
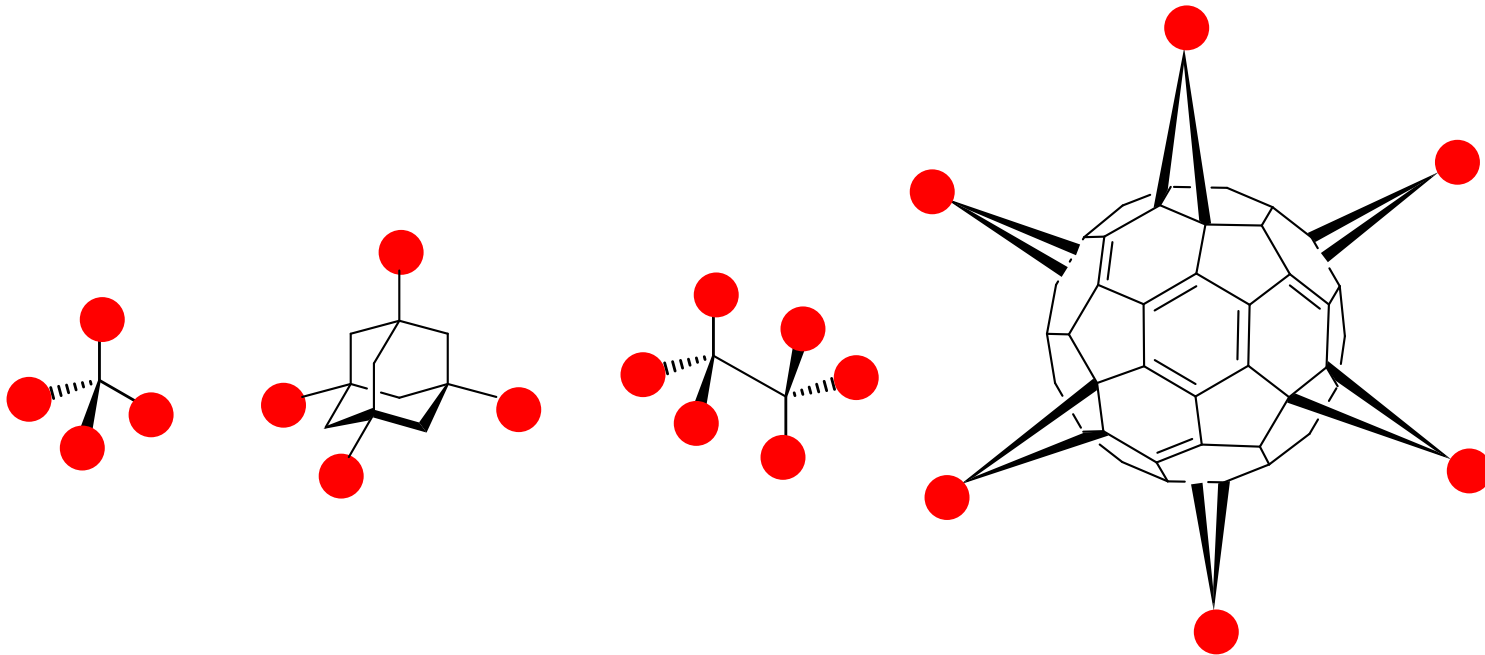


M. Lang



S. Grosjean

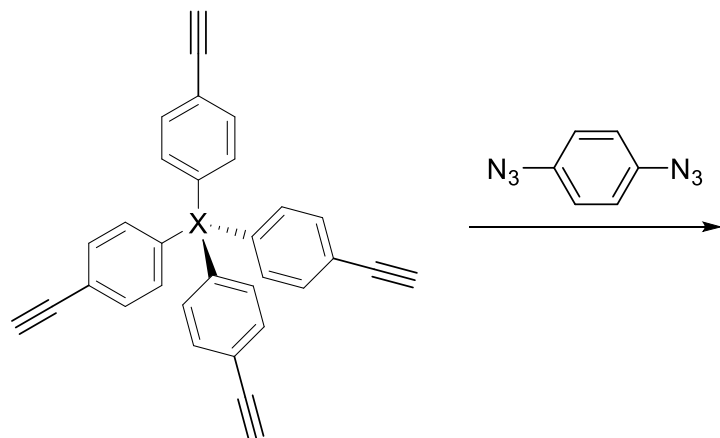
# Tetra- and (pseudo)octahedral organic cores





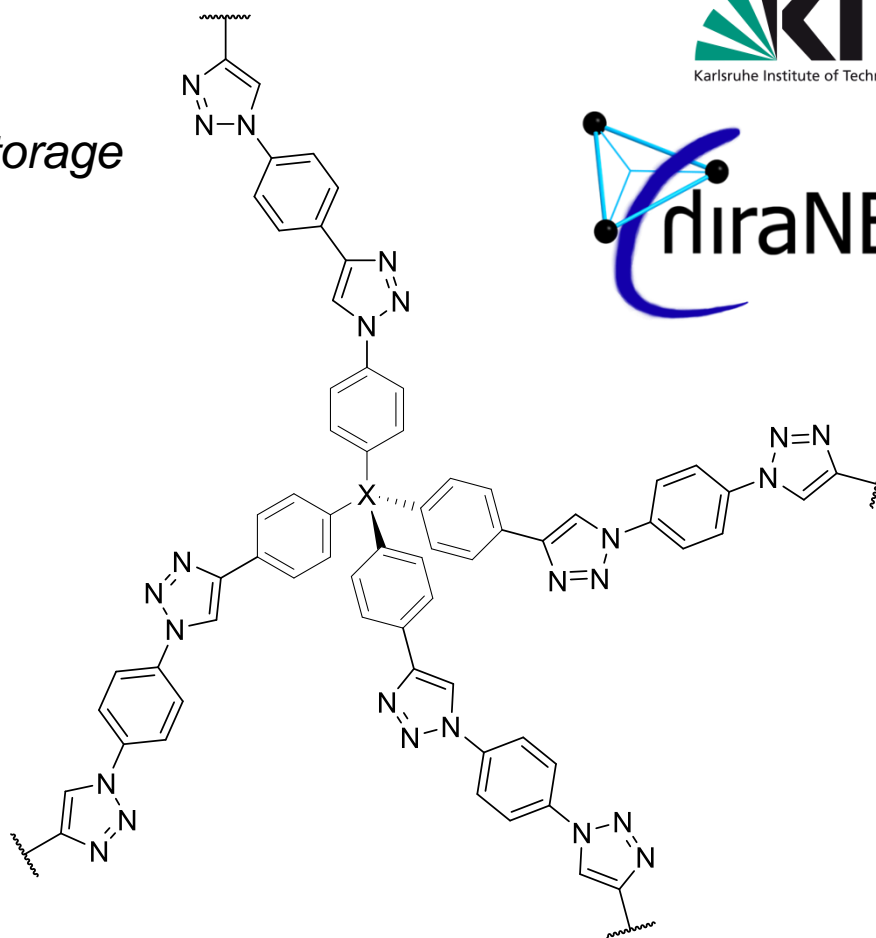
# Tetrahedral cores

Materials for gas storage  
and **catalysis**



X = C or adamantyl-1,3,5,7-tetrayl

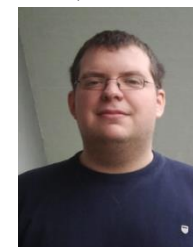
*New J. Chem.* **2011**, 15, 1577–1581  
*Chem. Mater.* **2010**, 22, 5964–5972  
(Eglington etc.)



A. Schade



I. Wessely

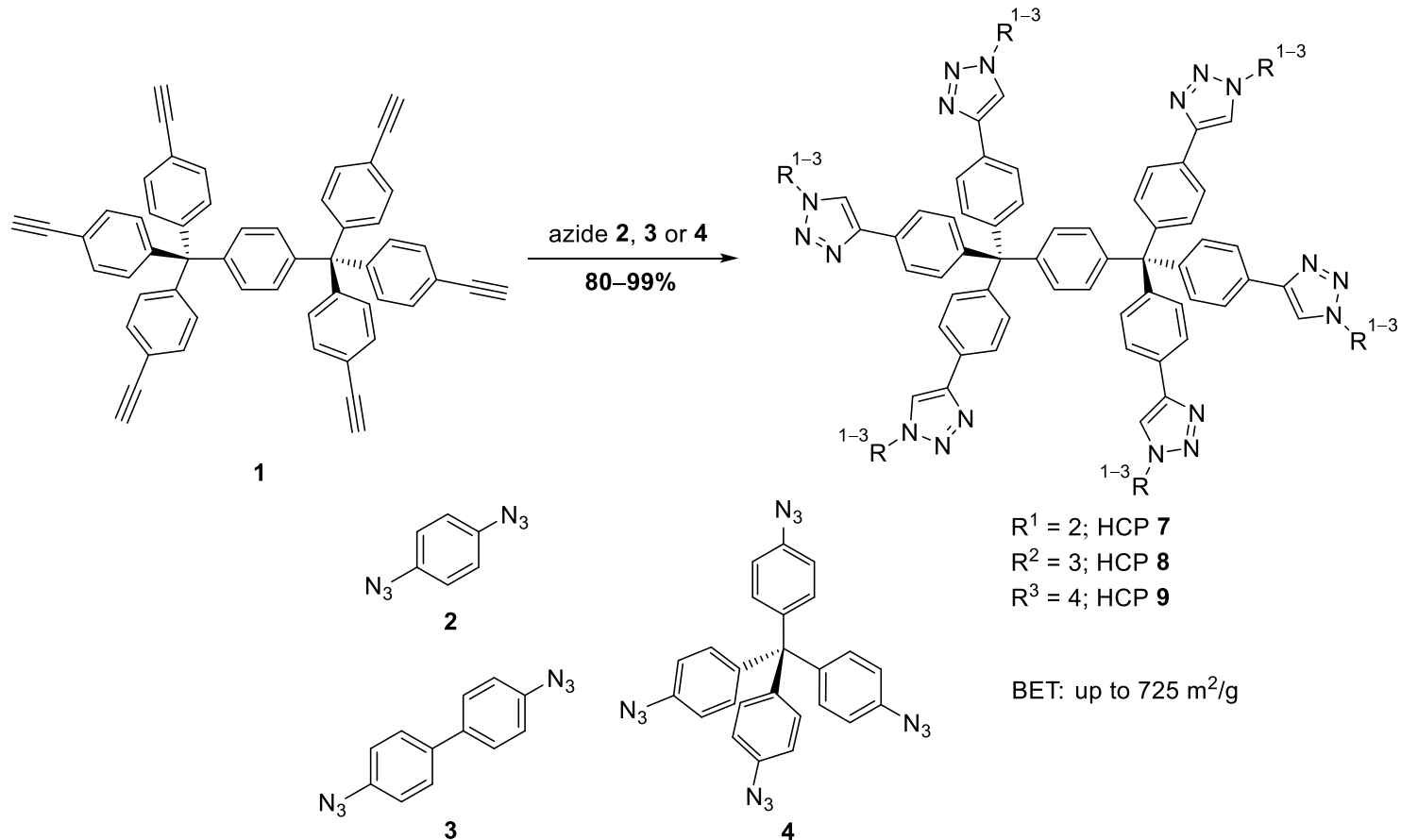


M. Lang



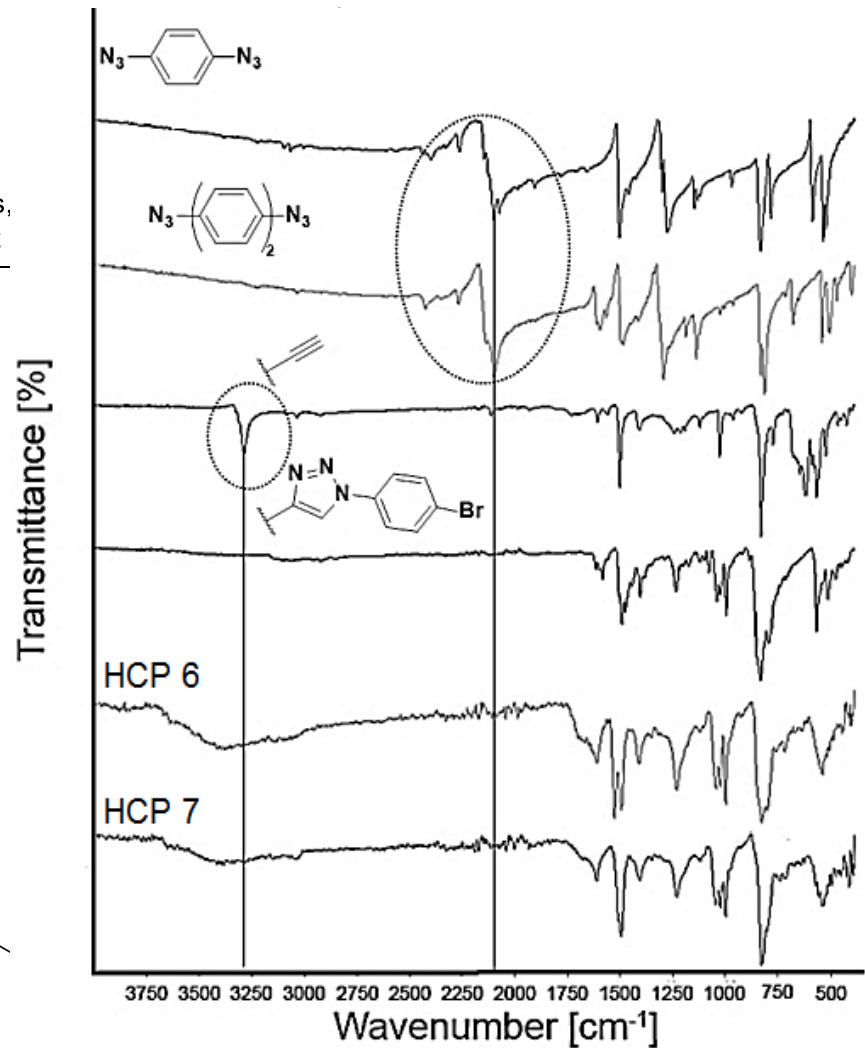
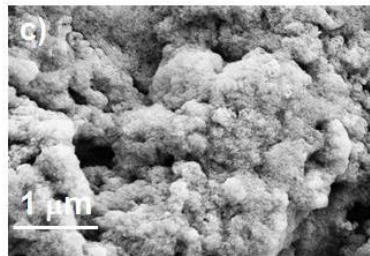
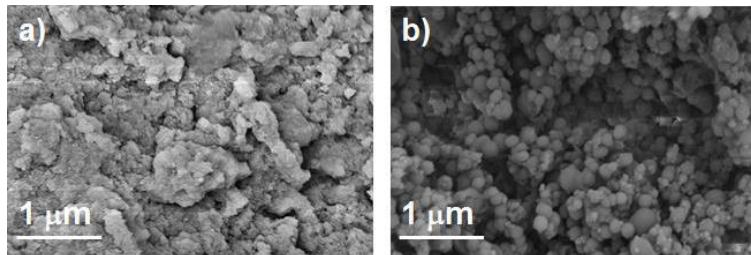
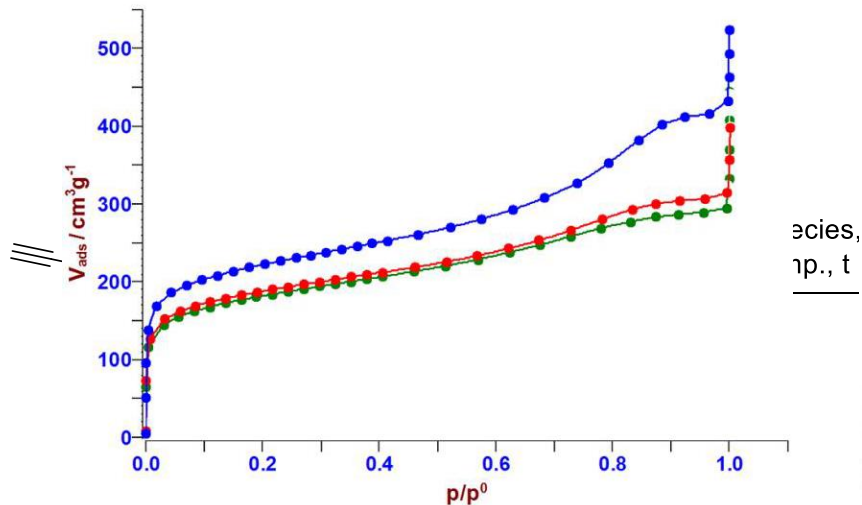
S. Grosjean

# Hexaethynyl bistritylbenzenes as building blocks



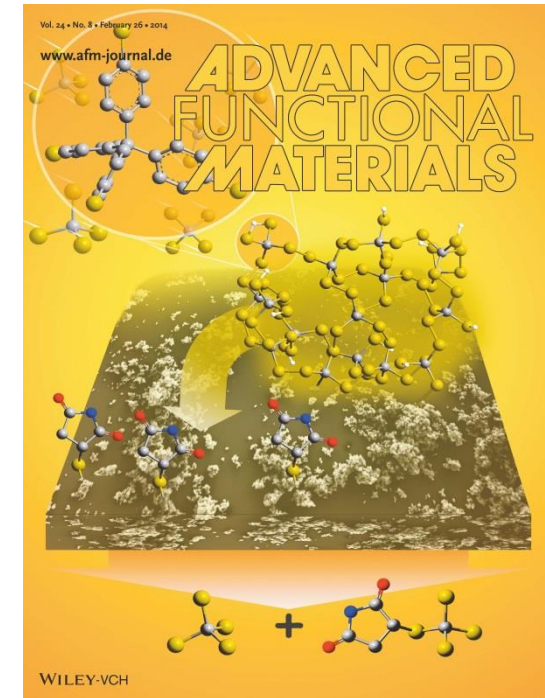
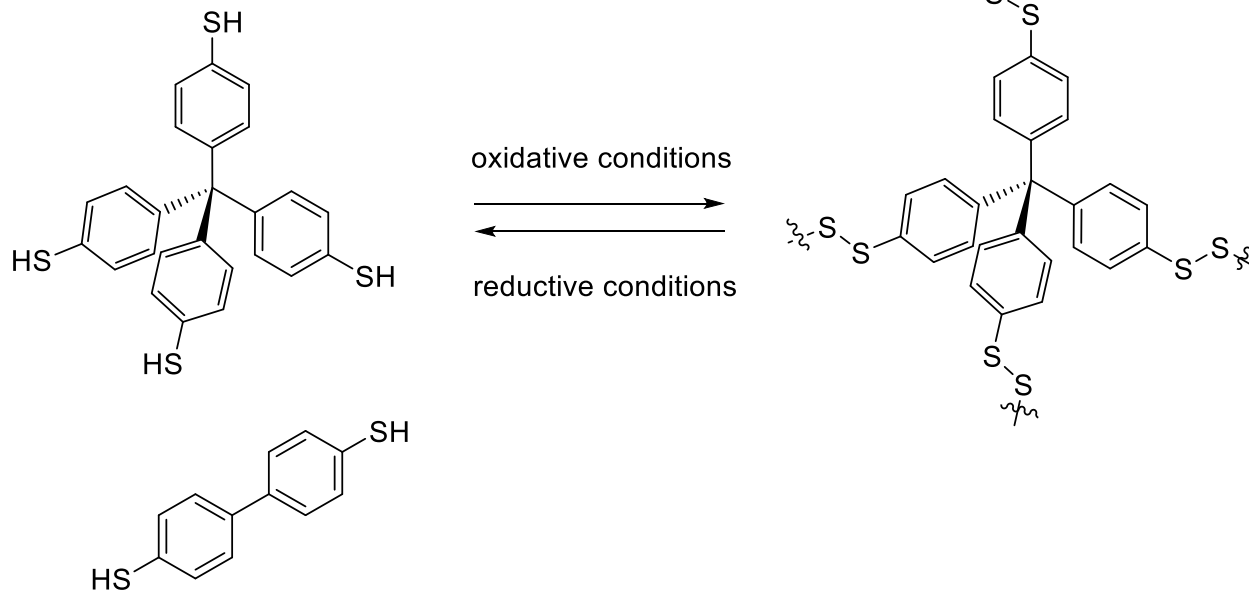
A. Schade et al., *ChemPlusChem* 2014

# Hexaethynyl bistritylbenzenes as building blocks



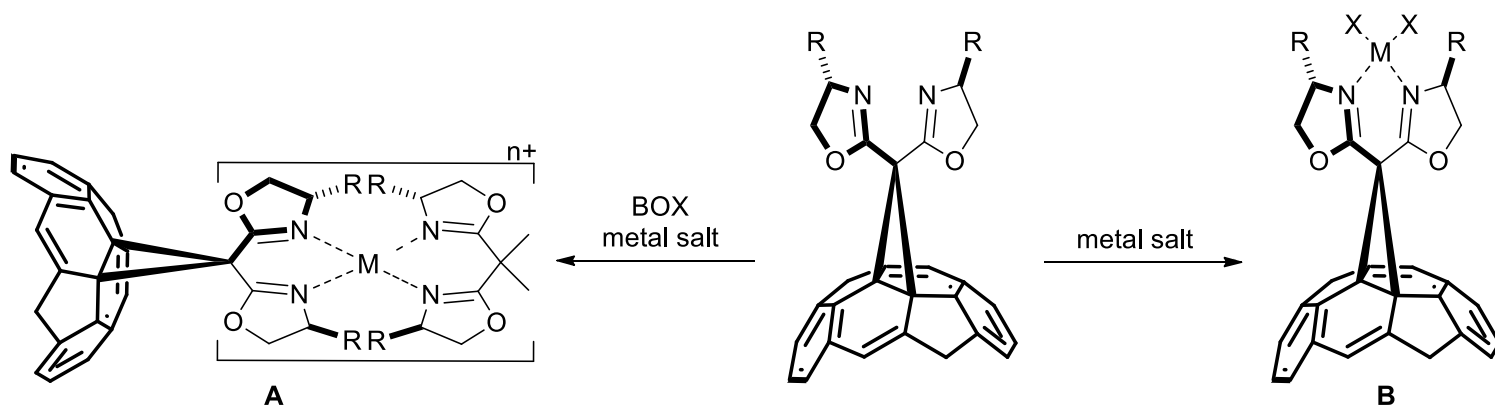
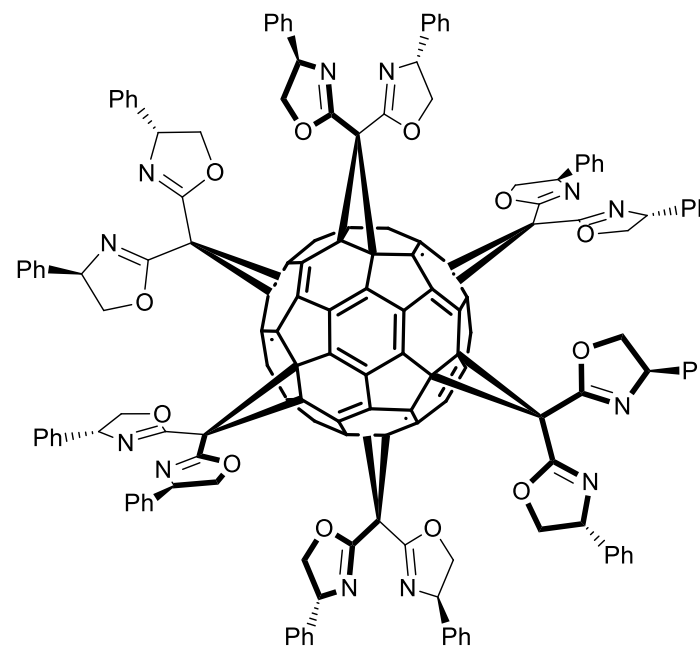
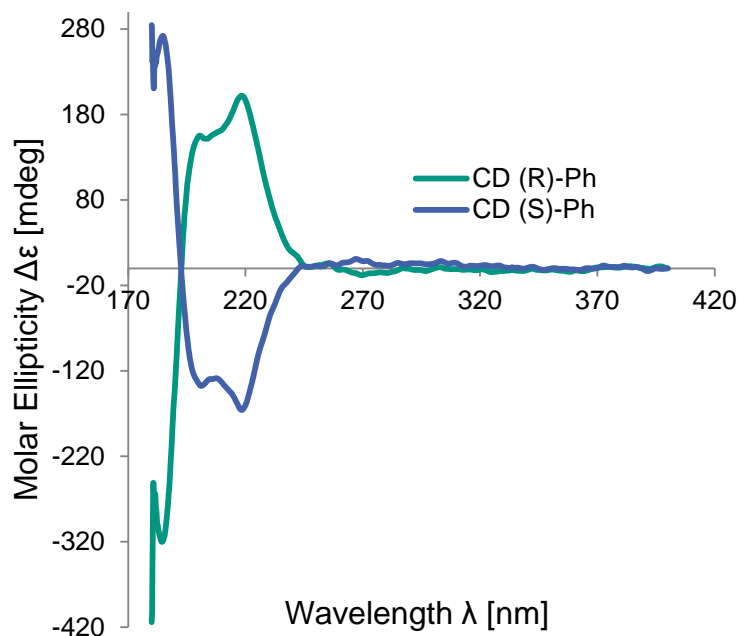
A. Schade et al., ChemPlusChem 2014

# Reversible COFs by Disulfides

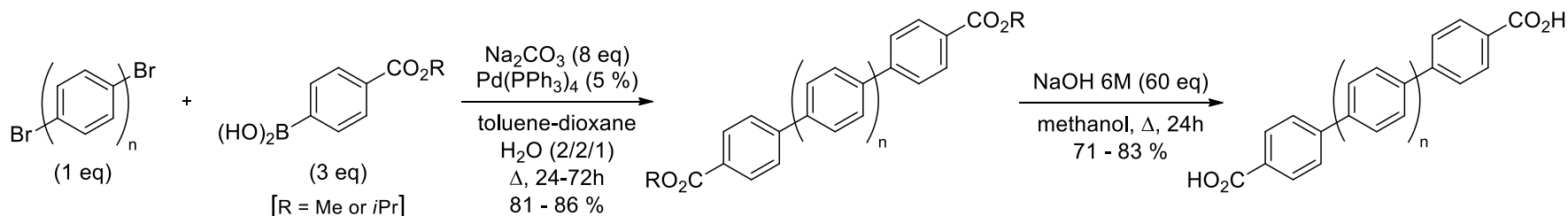


*L. Monnerau et al., Adv Funct Mat 2014*

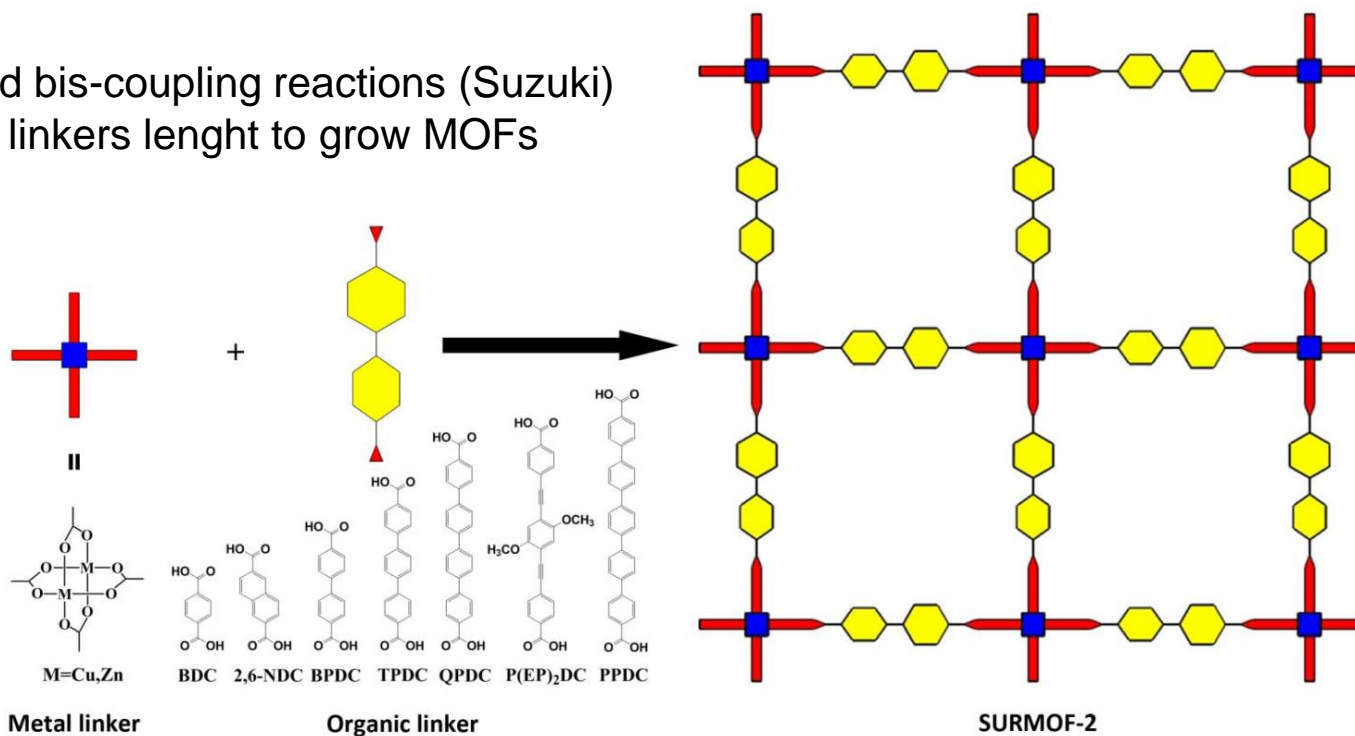
# Hexakis Fullerene Adducts of Bisoxazolines Asymmetric Catalysts for Nanofiltration



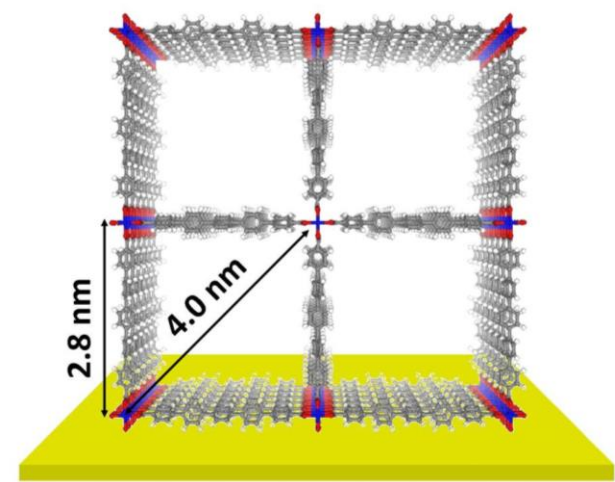
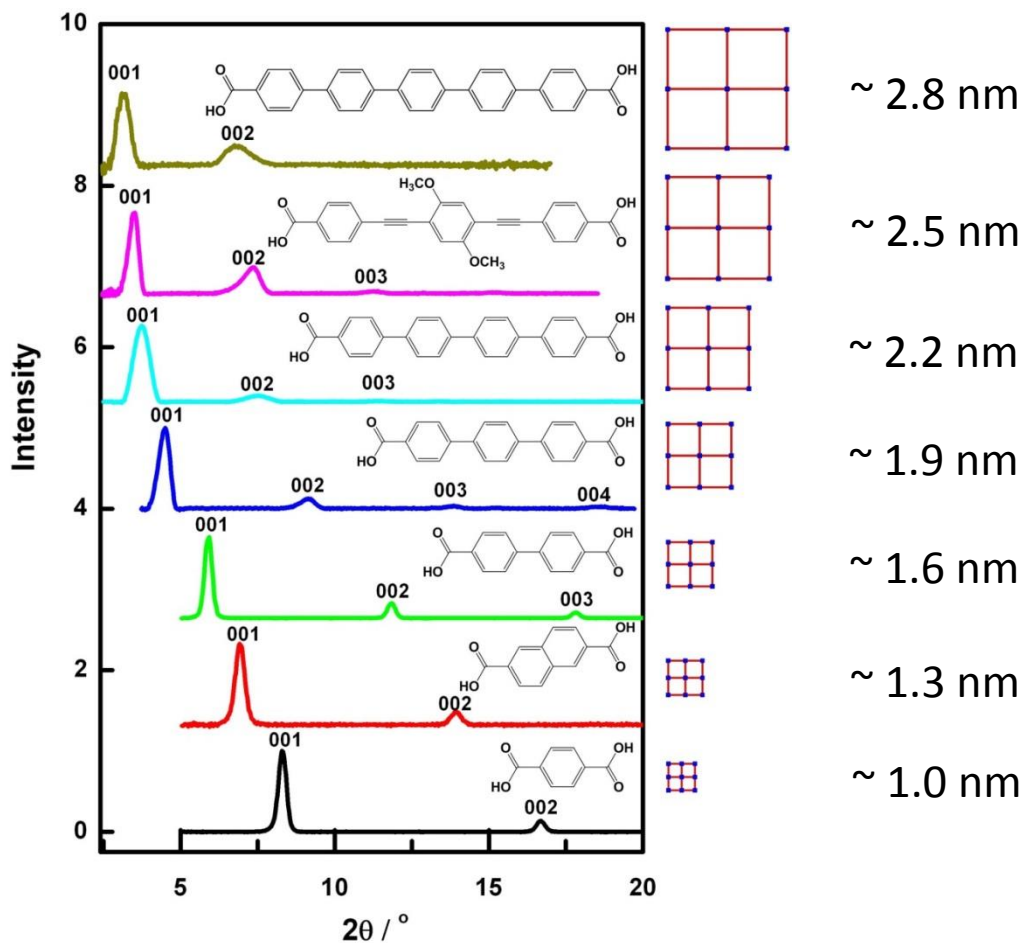
## Surface-mounted Metal-Organic Frameworks (SURMOFs)



Pd-assisted bis-coupling reactions (Suzuki)  
Increasing linker length to grow MOFs



## Surface-mounted Metal-Organic Frameworks (SURMOFs)



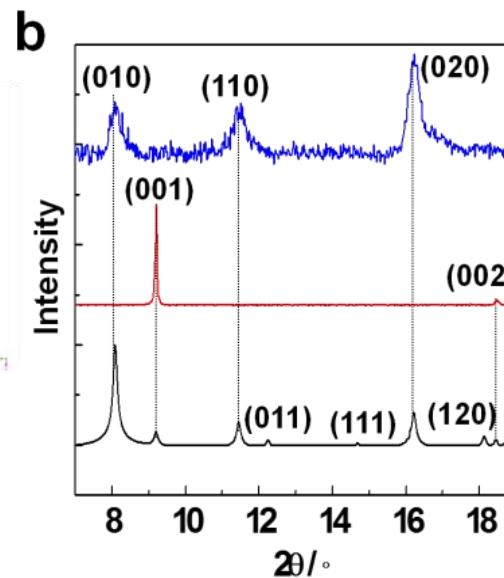
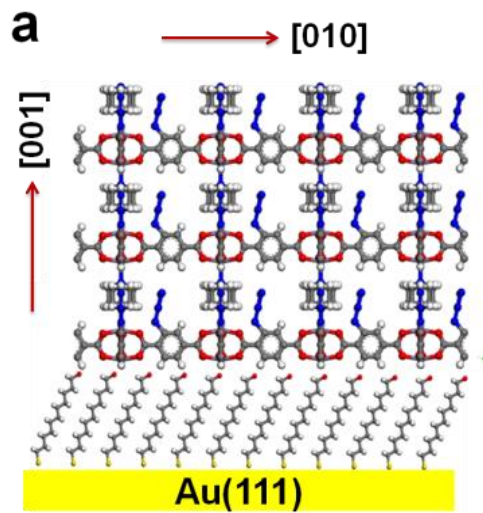
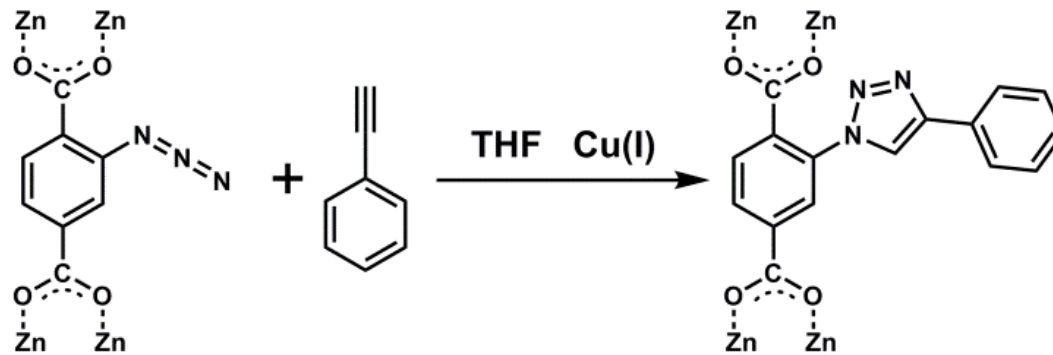
- Increasing linkers length
- Increasing pore size in MOFs

Sci. Rep. 2012, 2, 921.

# SURMOFs

## Postfunctionalization I

- Post-Synthetic Modification of MOFs: Cu(I)-catalyzed Huisgen Cyloaddition



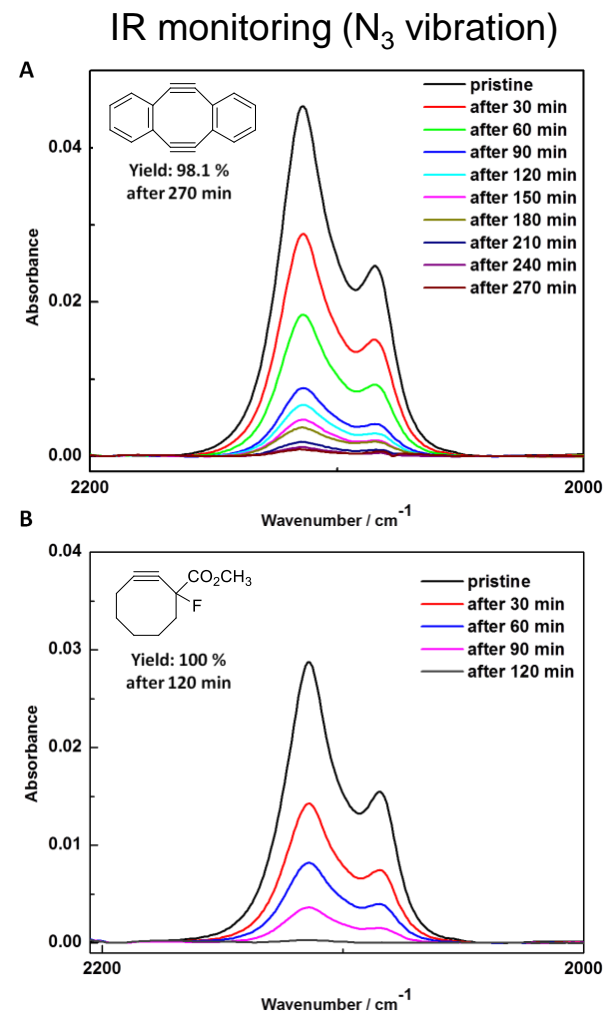
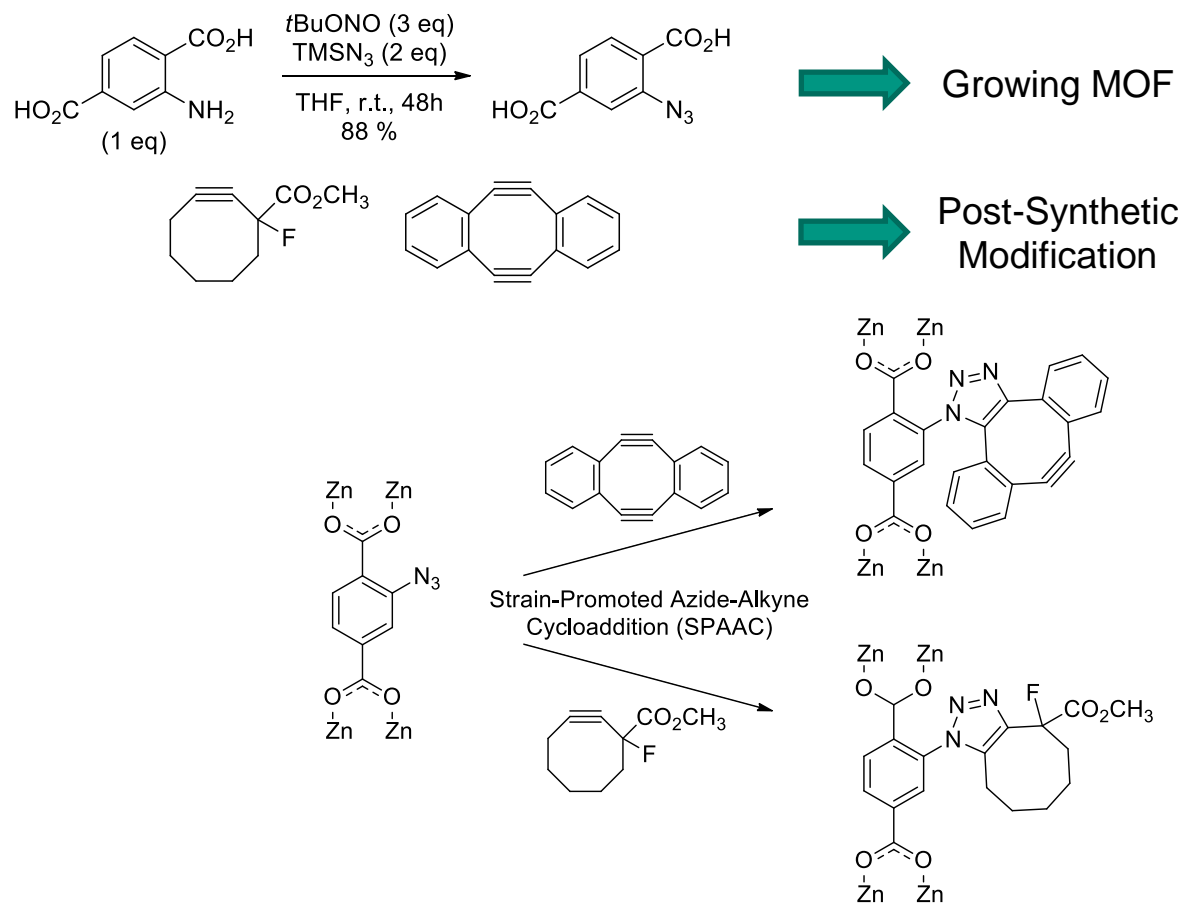
Langmuir **2013**, 29, 15958-15962.



# SURMOFs

## Postfunctionalization II

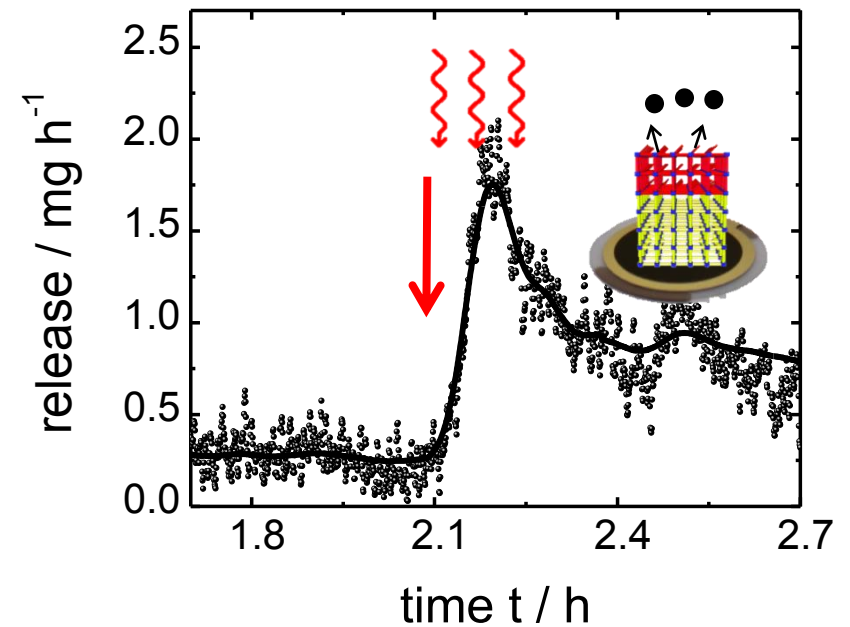
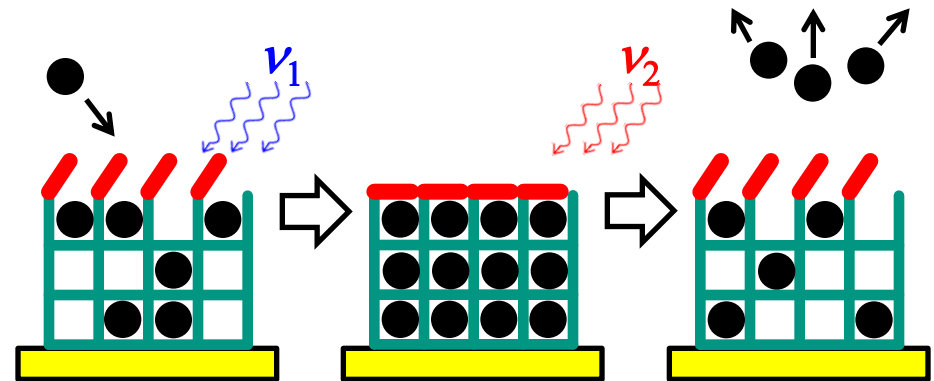
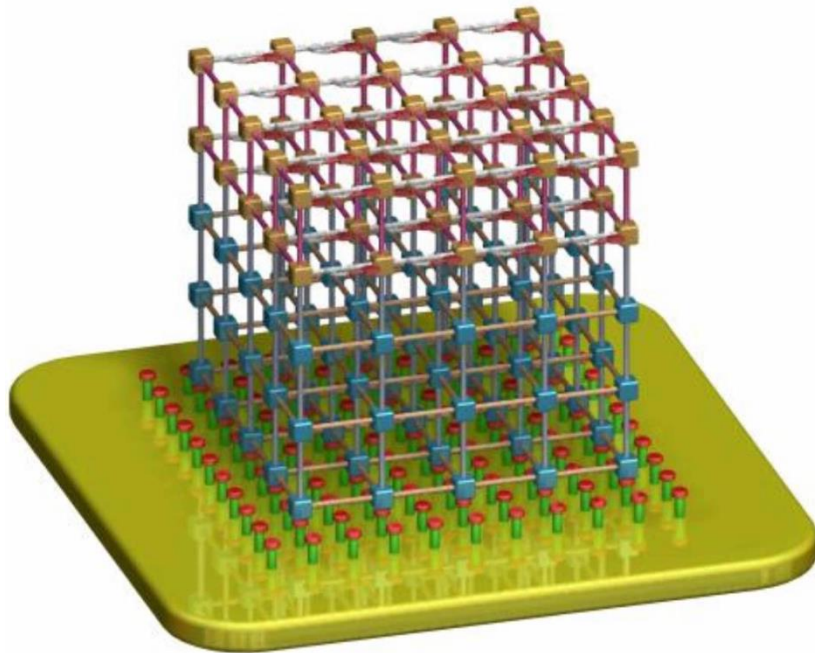
### Post-Synthetic Modification of MOFs: Strain Promoted Azide-Alkyne Cycloaddition (SPAAC)



Langmuir 2013, 29, 15958-15962.

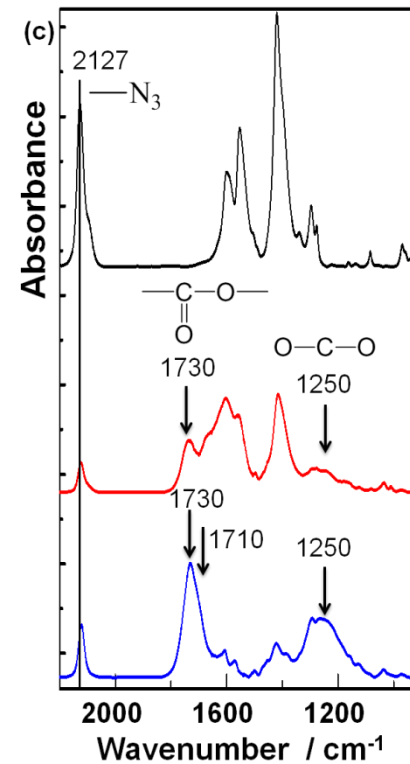
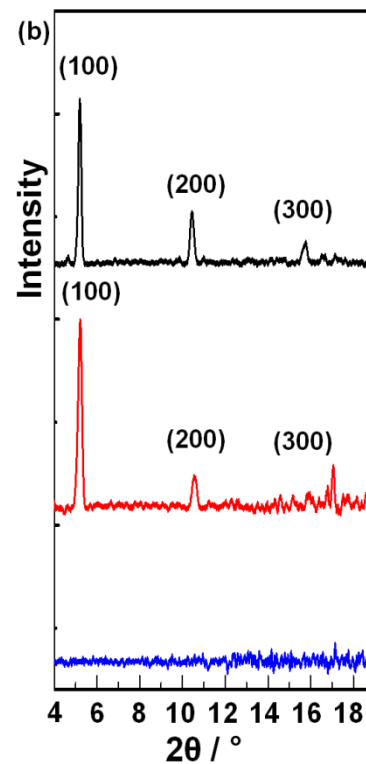
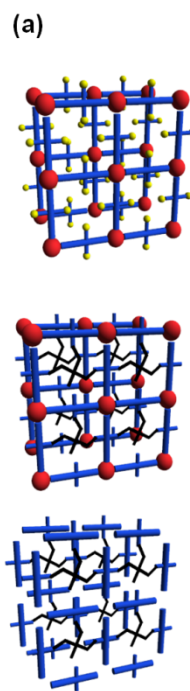
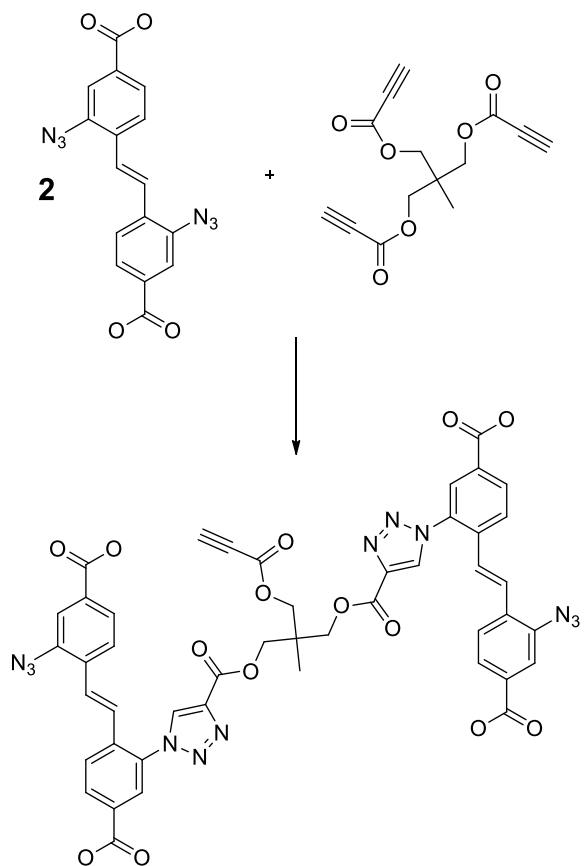
# SURMOFs

## Switchable Molecular Container using Azobenzenes



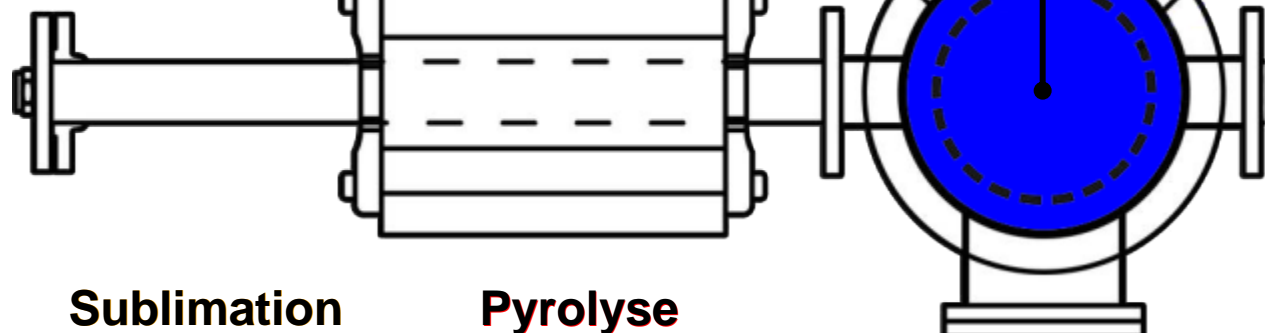
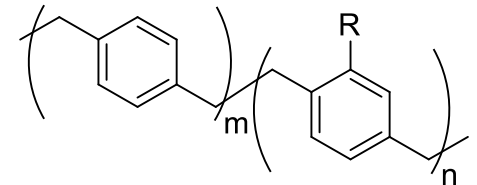
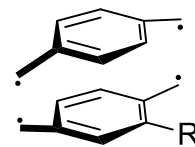
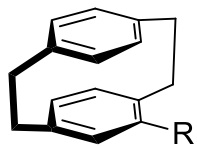
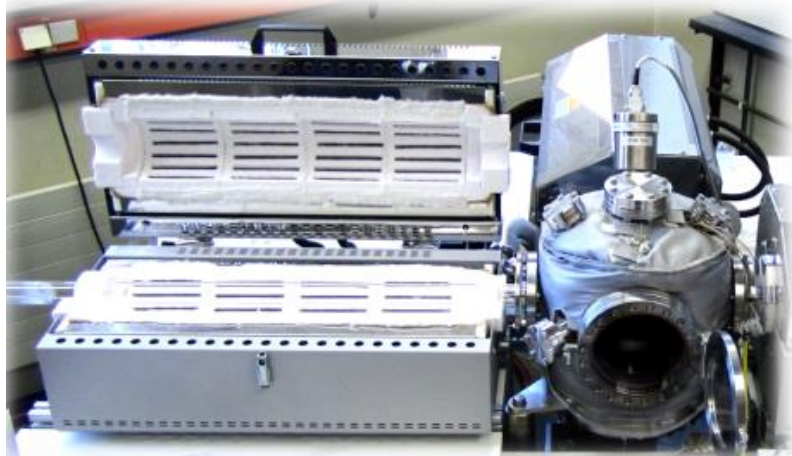
ACS Nano 2014, 8, 1463.

## From SURMOFs to Surface Structured Gels (SURGELS)



*J. Am. Chem. Soc.* **2014**, *136*, 8-11.

# Functionalized Polymers by Chemical Vapour Deposition Method



**Sublimation**

**Pyrolyse**

**Abscheidungs-  
kammer**

90 - 120 °C

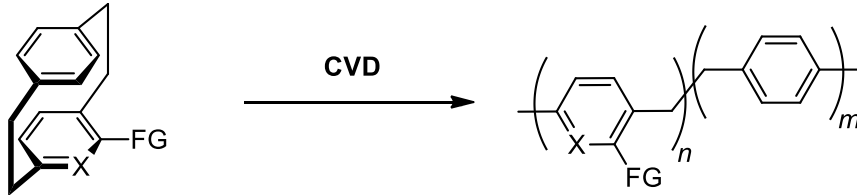
500 - 800 °C

-40 - 60 °C

H-Y. Chen, J. Lahann, *Langmuir* **2011**, 27, 34–48.

# Functionalised Paracyclophanes and Pyridinophanes for CVD-Polymerization

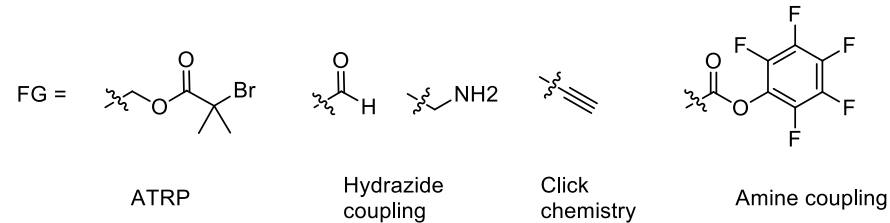
## CVD-Process



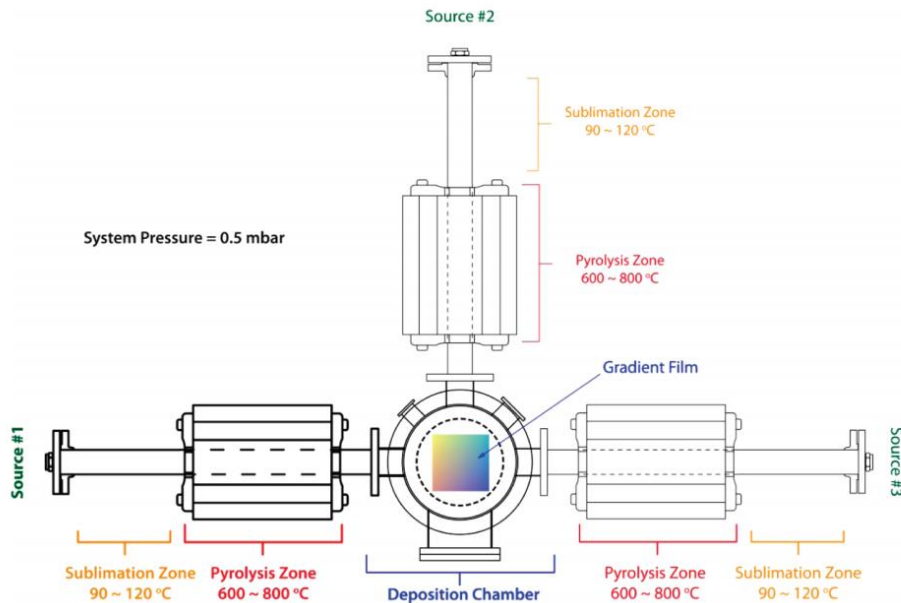
X = CH: PCP-FG  
X = N: PYP-FG

X = CH: PPX-FG  
X = N: PPLX-FG

## Selection of functional groups used in CVD and possible subsequent chemistry



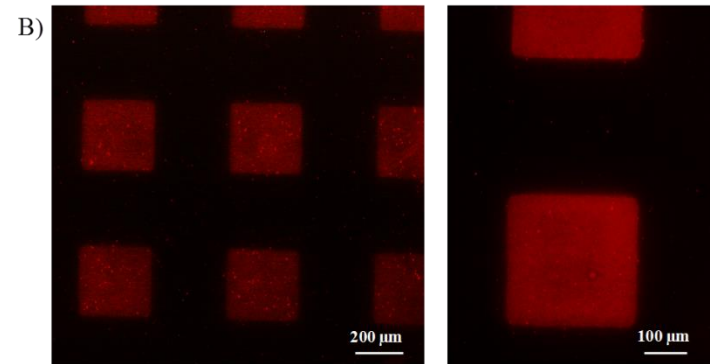
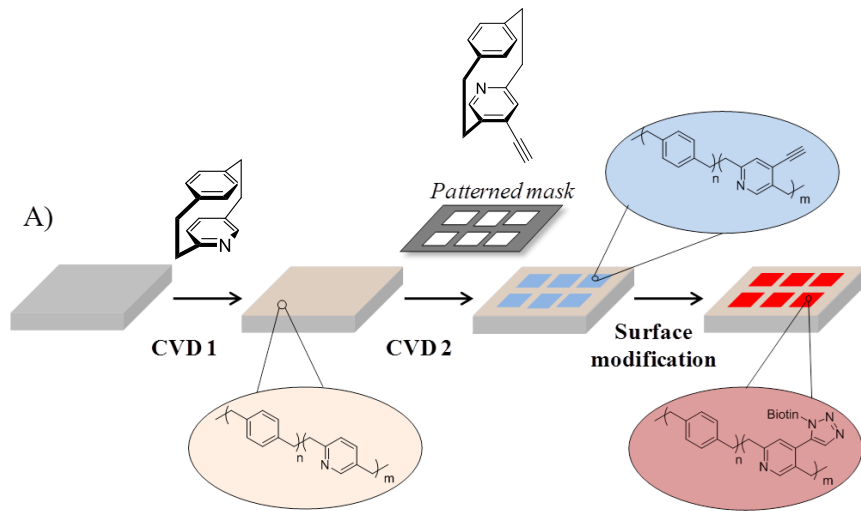
## Three-source CVD-system



## Applications of CVD coatings

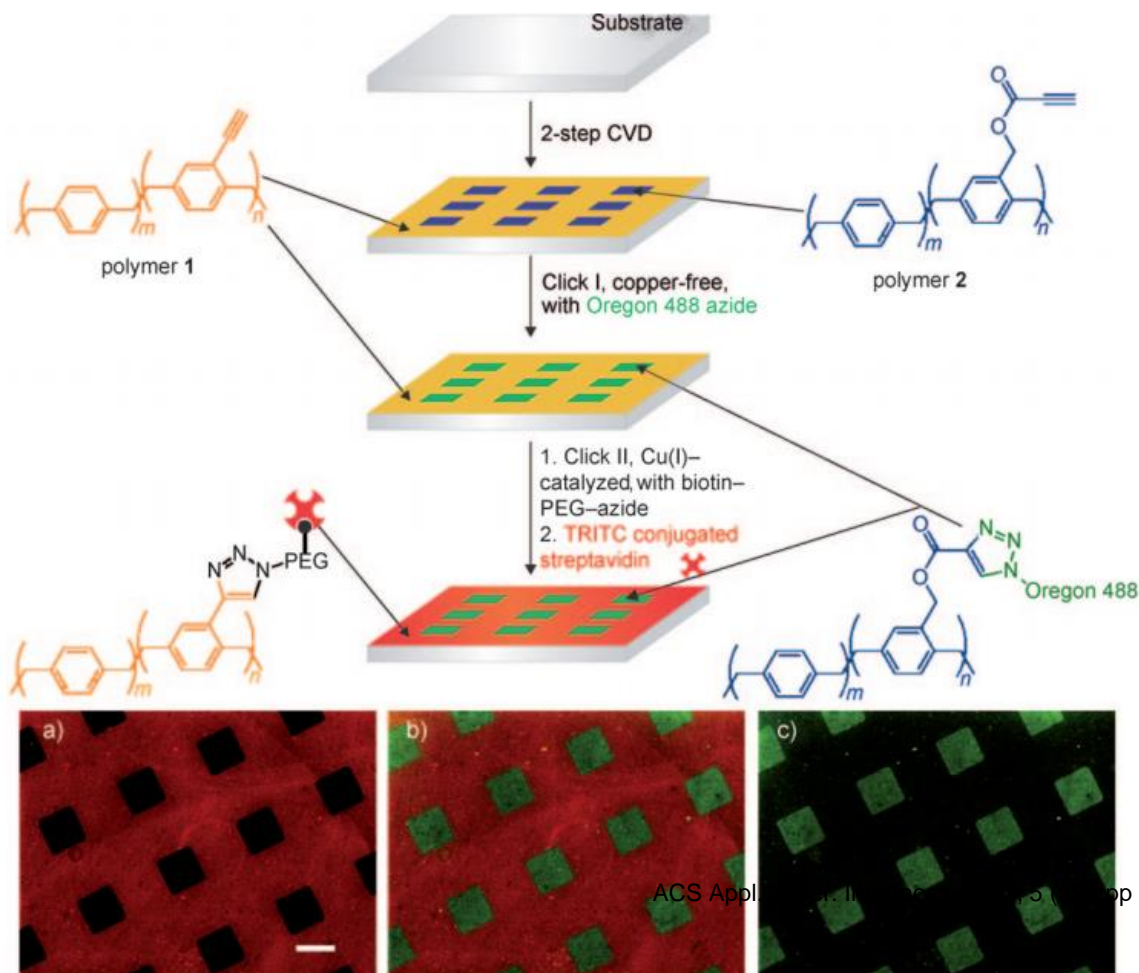
- Immobilization of cells
- Thermal bonding: Sensor technology
- Photonics
- **Metal complexes on surface**
- **Catalysis**

# New reactive coatings based on functional pyridinophanes



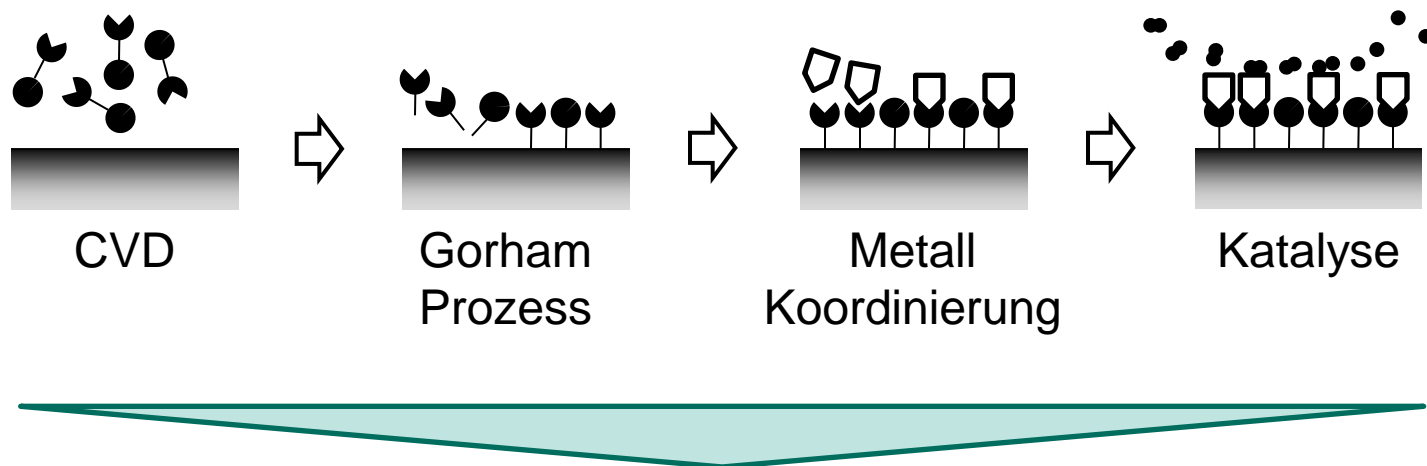
A) Schematic representation of the protocol used for biotin immobilization on the functional coating and B) fluorescence images after streptavidin incubation.

# Orthogonal “Double-Click” Chemistry and Co-immobilization of Molecules on CVD films

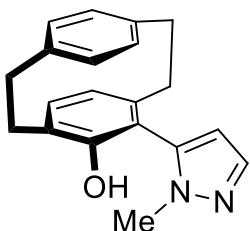


Deng, X., Friedmann, C. and Lahann, J. (2011), Bio-orthogonal “Double-Click” Chemistry Based on Multifunctional Coatings. *Angew. Chem. Int. Ed.*, 50: 6522–6526

# Functionalized Polymers for Catalysis by Chemical Vapour Deposition



## 7 [2.2]Paracyclophan-Derivate in 25 Experimenten getestet

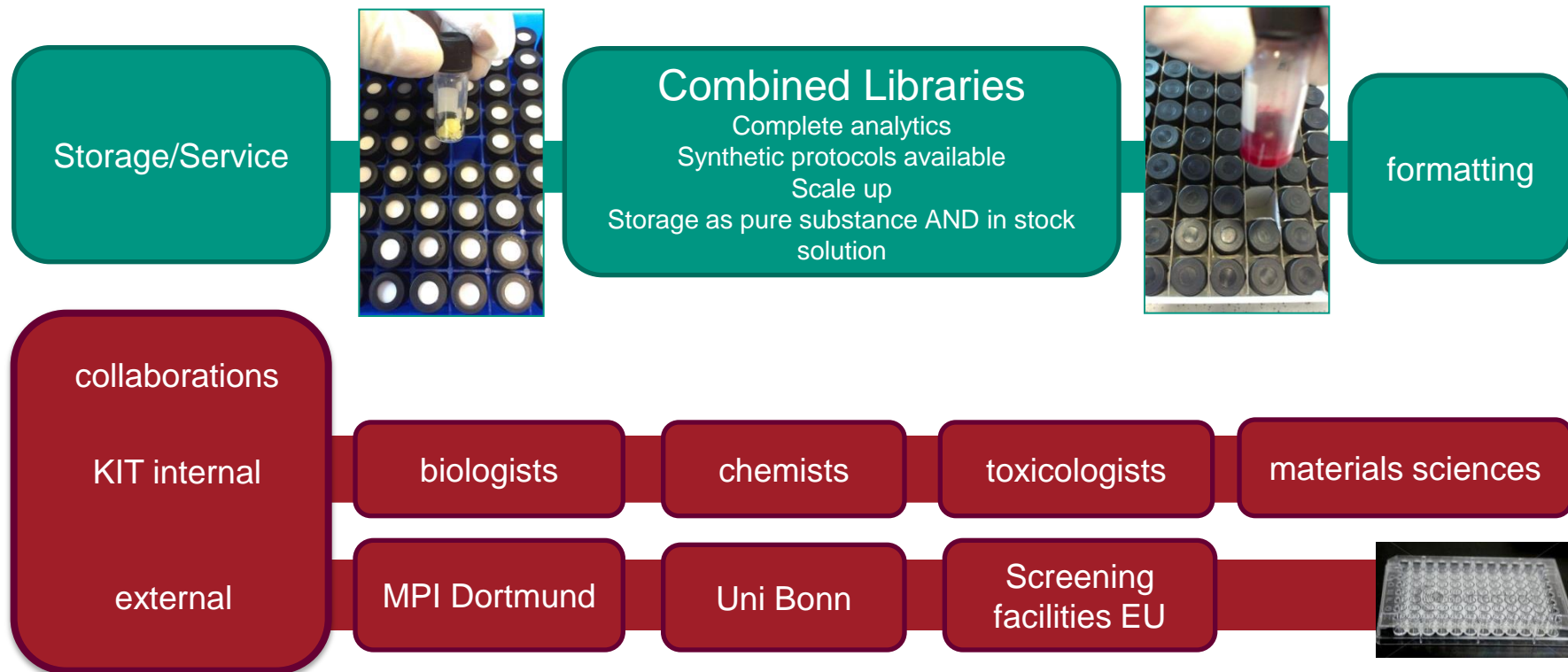


<b>T (Pyrolyse)</b>	660 °C	660 °C	750 °C
<b>T (Ablagerung)</b>	15 °C	15 °C	12 °C
<b>Abscheidungsrate</b>	0.5 Å/s	1.0 Å/s	1.0 Å/s
<b>Monomermasse</b>	23 mg	25 mg	55 mg
<b>Ellipsometer</b>	24.8 +/- 0.1 nm	35.6 +/- 0.1 nm	91.7 +/- 0.2 nm

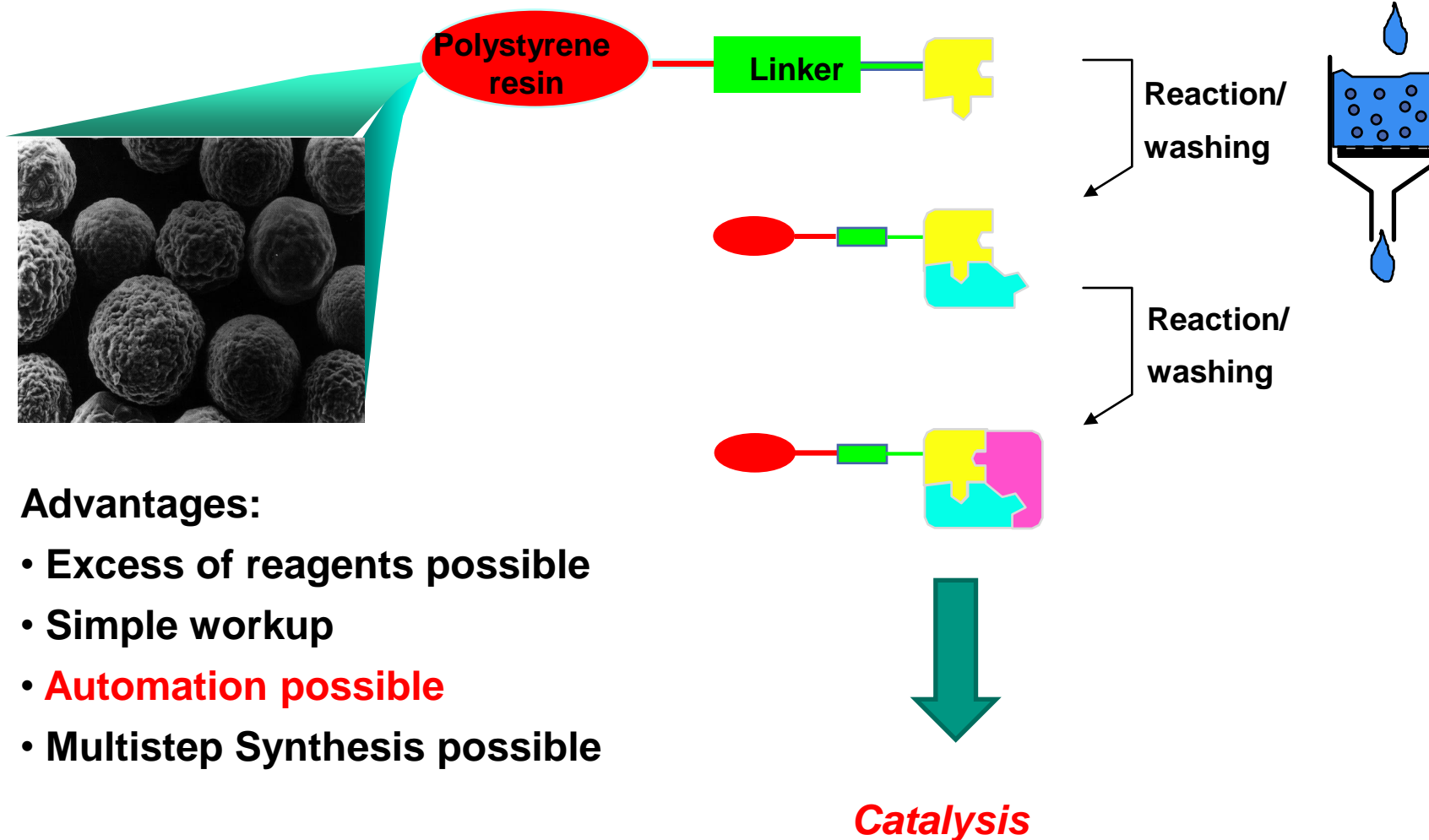
F. Bally, M. Busch, S. Bräse, J. Lahann, *unveröffentlicht*.



# Complat as Service Platform for Combinatorial Chemistry at the KIT



# Solid Phase-Bound Catalysts for Combinatorial Catalysis

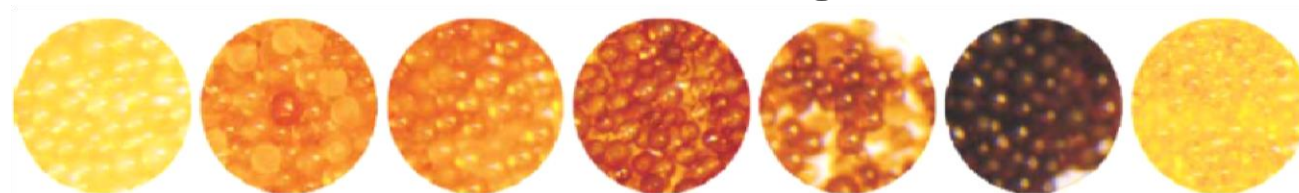
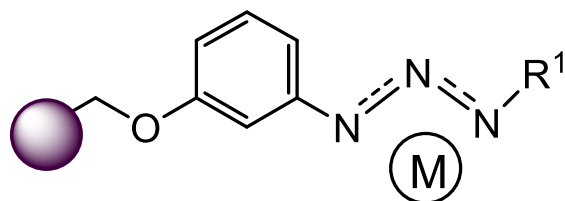


## Advantages:

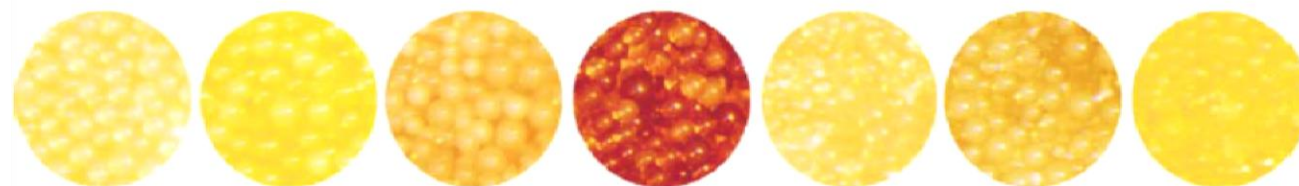
- Excess of reagents possible
- Simple workup
- **Automation possible**
- Multistep Synthesis possible

# Triazenes as Linkers for Metal Ions

## Metal-Binding



R = indanyl



R = ephedrinyll

Blank    Co    Cu    Fe    Pd    Ti    Zr

BMCL 2002 (with N. Leadbeater)

S. Bräse, S. Dahmen, *Chem. Eur. J.* **2000**, 5, 1899-1905.

S. Bräse, *Acc. Chem. Res.* **2004**, 37, 805-815.

*Angew. Chem. Int. Ed.* **1998**, 37, 3413-3415

*Angew. Chem. Int. Ed.* **1999**, 38, 1071-1073

*Angew. Chem. Int. Ed.* **1999**, 38, 3669-3672

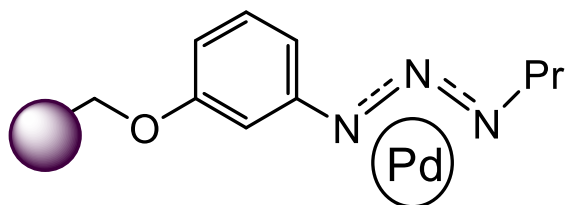
*Angew. Chem. Int. Ed.* **2008**, 47, 8120-8122

*Angew. Chem. Int. Ed.* **2010**, 49, 5986-5988

*Angew. Chem. Int. Ed.* **2011**, 50, 11533-11535

*Angew. Chem. Int. Ed.* **2012**, 51, 3713-3715

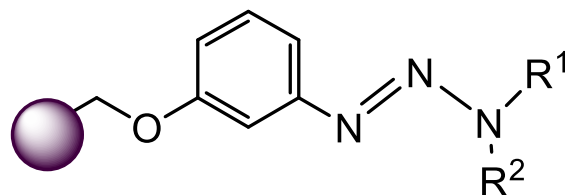
# Triazenes as Linkers for Metal Ions Catalysis



very low leaching

Sonogashira

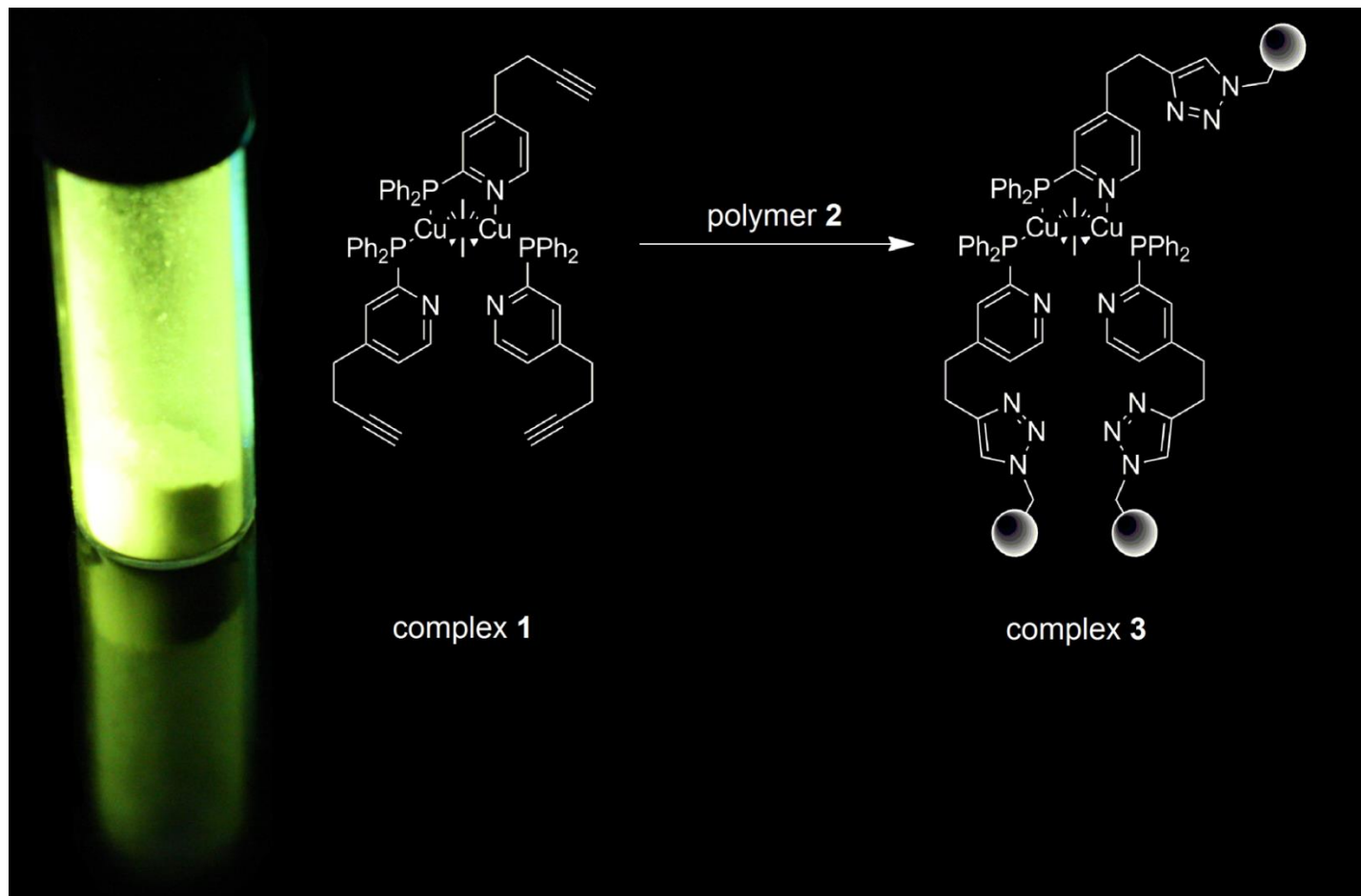
Suzuki



Asymmetric 1,2-additions

*BMCL 2002* (with N. Leadbeater)

# Concepts for immobilization of Metal Complexes

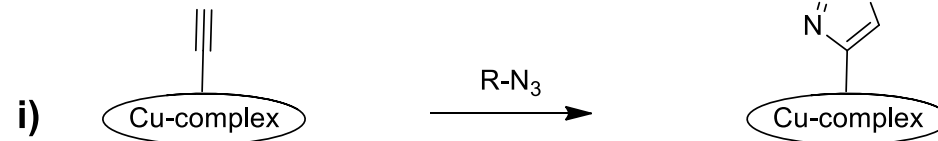


Volz, D.; Baumann, T.; Flügge, H.; Mydlak, M.; Grab, T.; Bächle, M.; Barner-Kowollik, C.; Bräse, S.  
*J. Mater. Chem.* **2012**, *22*, 20786.

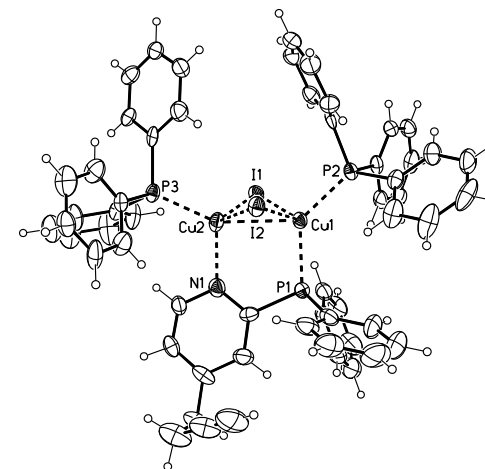
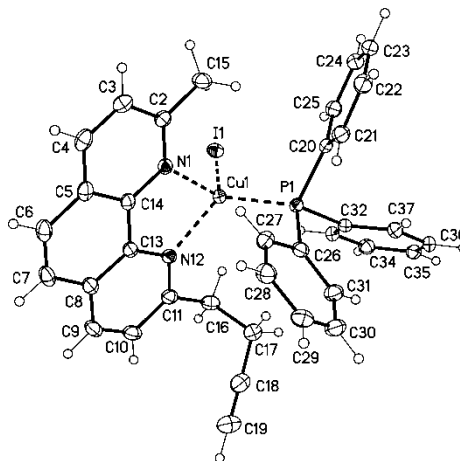
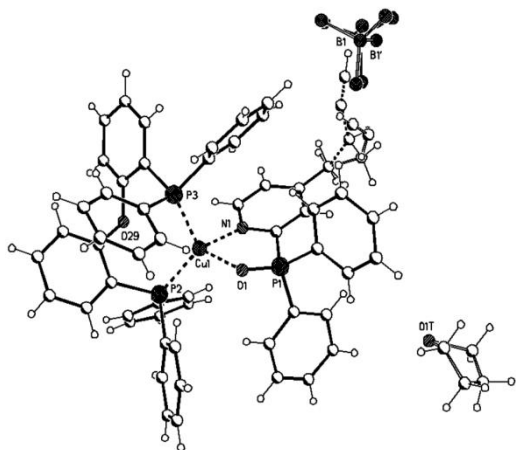
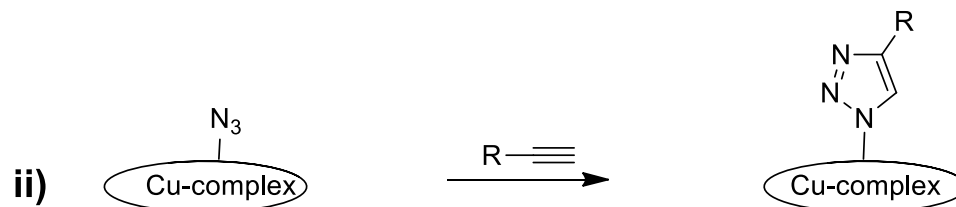
# Self-catalyzed Click-reaction of alkyne-Cu(I) complexes

## Concept

1) functionalize Cu(I) complex (mononuclear, binuclear..)

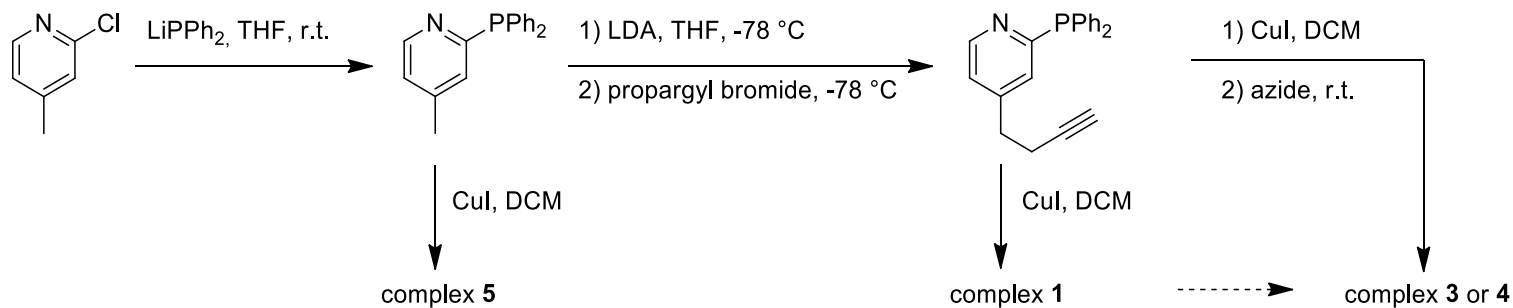
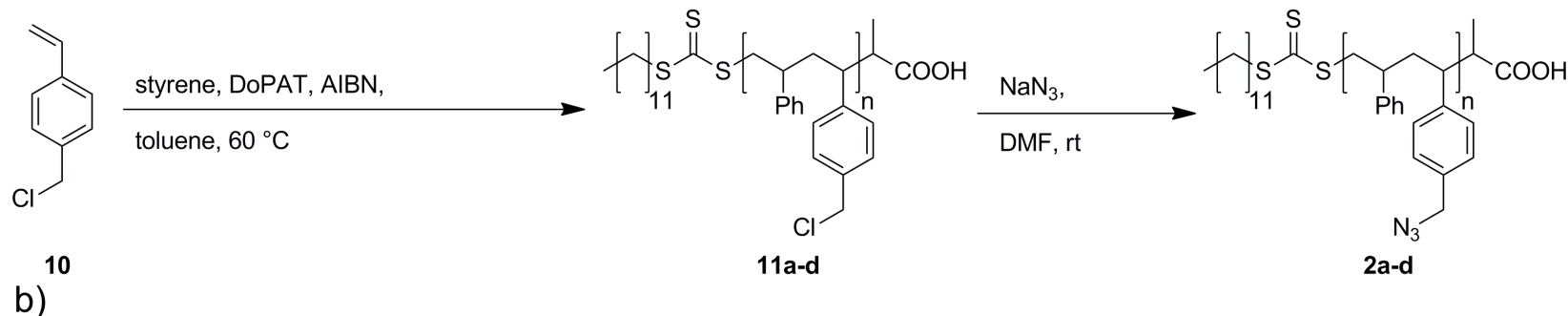


2) react with small molecule/polymer/functionalized surface



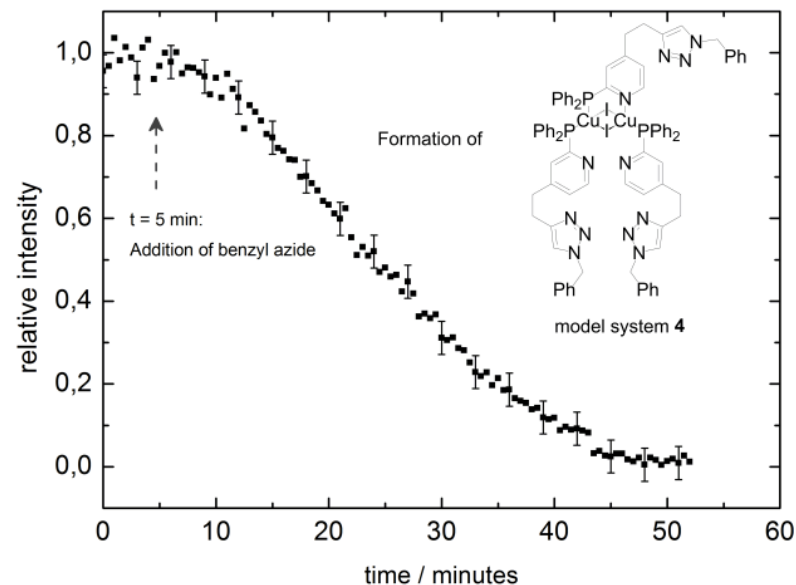
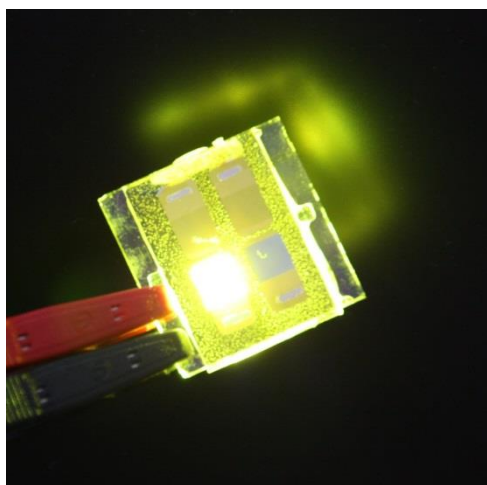
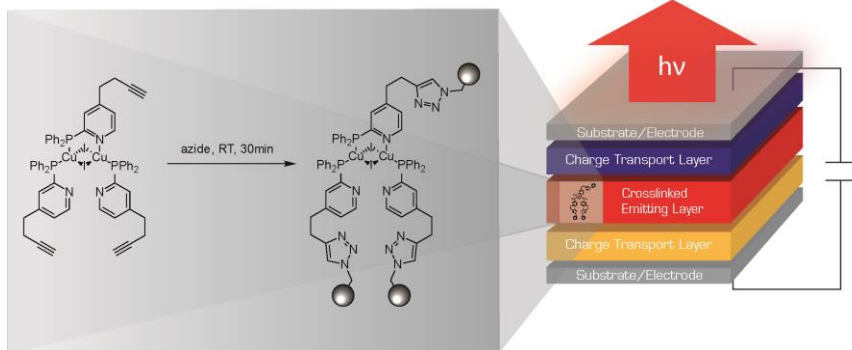
D. Volz, M. Wallesch, S. Bräse, *unpublished results*

# Self-catalyzed Click-reaction of alkyne-Cu(I) complexes



Volz, D.; Baumann, T.; Flügge, H.; Mydlak, M.; Grab, T.; Bächle, M.; Barner-Kowollik, C.; Bräse, S.  
*J. Mater. Chem.* **2012**, *22*, 20786.

# Self-catalyzed Click-reaction of alkyne-Cu(I) complexes



reaction of complex with benzyl azide has been monitored with FT-IR spectroscopy

concept has been demonstrated successfully in a crosslinked OLED device

Volz, D.; Baumann, T.; Flügge, H.; Mydlak, M.; Grab, T.; Bächle, M.; Barner-Kowollik, C.; Bräse, S. *J. Mater. Chem.* **2012**, *22*, 20786.



# Today's menu

- **Aperitif: Immobilization of Homogeneous Catalysts**
- **Amuse-bouche: Asymmetric Catalysis**
- **Entrée: Immobilization of Catalysts**
- **Main course: Combinatorial Chemistry**
- **Dessert: Diversity through Multifunctional Linker**
- **Cheese: Auto-Click Reactions**

# Bräse group: Current small molecule chemistry team

## Secalonic acids/Beticolin

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## Cannabinoids

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## Steroids/Wnt-signalling

Vanessa Koch  
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Valentin Beyer  
Dr. Alicia Dilmac

## Biointerfaces/CUBUSLab

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## KIT CombiChemistry Platform

Dr. Nicole Jung  
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Steven Susanto  
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## Sugars and more

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Mirella Wawryczyn  
Anna Fellingner  
Katharina Peschko  
Anna Hörner  
Anne Schneider  
Stephan Münch  
Alexander Braun

## Nano structures

Dr. Christin Bednarek  
Alexandra Schade  
Dr. Sylvain Grosjean  
Isabelle Wessely  
Mathias Lang  
Patrick Hodapp

## Metals and more

Larissa Bergmann  
Manuela Wallesch  
Angela Wandler  
Carolin Braun  
Ceylan Yildiz  
Eduard Spuling



FCI  
Alexander-von-Humboldt foundation  
DFG: TR 88, KSOP, CFN  
BASF  
Bayer AG

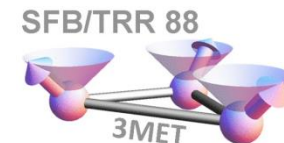


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